2011 Vol. 13, No. 8 2022–2025

Highly Enantioselective and Efficient Synthesis of Flavanones Including Pinostrobin through the Rhodium-Catalyzed Asymmetric 1,4-Addition

Toshinobu Korenaga,* Keigo Hayashi, Yusuke Akaki, Ryota Maenishi, and Takashi Sakai*

Division of Chemistry and Biochemistry, Graduate School of Natural Science and Technology, Okayama University, 3-1-1 Tsushima-naka, Okayama 700-8530, Japan

korenaga@cc.okayama-u.ac.jp

Received February 15, 2011

ABSTRACT [RhOH(cod)]₂ / (S)-MeO-F₁₂-BIPHEP (4) (0.5-3% Rh; 4/Rh = 1.0) Toluene / H₂O or CH₂Cl₂ / H₂O (S)-1 0-40 °C, 0.7-3 h > 80% yield (up to 95%) ≥ 99% ee

An efficient synthesis of bioactive chiral flavanones (1) was achieved through the Rh-catalyzed asymmetric 1,4-addition of arylboronic acid to chromone. The reaction in toluene proceeded smoothly at room temperature in the presence of 0.5% Rh catalyst with electron-poor chiral diphosphine MeO- F_{12} -BIPHEP. In this reaction, the 1,2-addition to (S)-1 frequently occurred to yield (2S,4R)-2,4-diaryl-4-chromanol as a byproduct, which could be reduced by changing the reaction solvent to CH_2Cl_2 to deactivate the Rh catalyst (3% required).

Flavanone (1a) and its derivatives (1) are found in many natural products that show biological activities¹ such as antioxidant,² antitumoral,³ and antiproliferative properties.⁴ However, the optically pure 1 had been prepared by optical resolution of racemic 1 or diastereoselective synthesis from a chiral material, and the efficient synthesis of 1 by asymmetric catalysis had not been known until recently. In 2007, Scheidt et al. reported the catalytic

enantioselective synthesis of 1^{5a} through an asymmetric intramolecular oxa-Michael addition and decarboxylation by using a chiral organic^{5a,b} or transition-metal catalyst.^{5c} Although this method is effective for the synthesis of 1, the asymmetric induction of an aryl group into commercially available chromone 2a would be a more convenient and efficient method to synthesize 1. In particular, the rhodium-catalyzed asymmetric 1,4-addition of arylboronic

^{(1) (}a) Harborne, J Nat. Prod. Rep 1995, 12, 639. (b) Chang, L. C.; Kinghorn, A. D. In Bioactive Compounds from Natural Sources: Isolation, Characterisation and Biological Properties; Tringali, C., Ed.; Taylor & Francis: London, 2001. (c) Flavonoids: Chemistry, Biochemistry and Applications; Andersen, Ø. M., Markham, K. R., Eds.; Taylor & Francis: London, 2006.

⁽²⁾ Pietta, P. J. Nat. Prod. 2000, 63, 1035.

⁽³⁾ Bracke, M. E.; Depypere, H. T.; Boterberg, T.; Van Marck, V. L.; Vennekens, K. M.; Vanluchene, E.; Nuytinck, M.; Serreyn, R.; Mareel, M. M. J. Natl. Cancer Inst. 1999, 91, 354.

⁽⁴⁾ Le Bail, J. C.; Varnat, F.; Nicolas, J. C.; Habrioux, G. Cancer Lett. 1998, 130, 209.

^{(5) (}a) Biddle, M. M.; Lin, M.; Scheidt, K. A. J. Am. Chem. Soc. 2007, 129, 3830. (b) Wang, H.-F.; Cui, H.-F.; Chai, Z.; Li, P.; Zheng, C.-W.; Yang, Y.-Q.; Zhao, G. Chem.—Eur. J. 2009, 15, 13299. (c) Wang, L.; Liu, X.; Dong, Z.; Fu, X.; Feng, X. Angew. Chem., Int. Ed. 2008, 47, 8670.

⁽⁶⁾ First report: Takaya, Y.; Ogasawara, M.; Hayashi, T.; Sakai, M.; Miyaura, N. J. Am. Chem. Soc. 1998, 120, 5579.

⁽⁷⁾ Reviews: (a) Miyaura, N. *Bull. Chem. Soc. Jpn.* **2008**, *81*, 1535. (b) Hayashi, T.; Yamasaki, K. *Chem. Rev.* **2003**, *103*, 2829. (c) Edwards, H. J.; Hargrave, J. D.; Penrose, S. D.; Frost, C. G. *Chem. Soc. Rev.* **2010**, *39*, 2093.

⁽⁸⁾ Recent examples: (a) Shintani, R.; Takeda, M.; Nishimura, T.; Hayashi, T. Angew. Chem., Int. Ed. 2010, 49, 3969. (b) Nishimura, T.; Wang, J.; Nagaosa, M.; Okamoto, K.; Shintani, R.; Kwong, F.-Y.; Yu, W.-Y.; Chan, A. S. C.; Hayashi, T. J. Am. Chem. Soc. 2010, 132, 464. (c) Chen, Q.-A.; Dong, X.; Chen, M.-W.; Wang, D.-S.; Zhou, Y.-G.; Li, Y.-X. Org. Lett. 2010, 12, 1928. (d) Luo, Y.; Carnell, A. J. Angew. Chem., Int. Ed. 2010, 49, 2750. (e) Lang, F.; Li, D.; Chen, J.; Chen, J.; Li, L.; Cun, L.; Zhu, J.; Deng, J.; Liao, J. Adv. Synth. Catal. 2010, 352, 843. (f) Brown, M. K.; Corey, E. J. Org. Lett. 2010, 12, 172. (g) Drinkel, E.; Briceno, A.; Dorta, R.; Dorta, R. Organometallics 2010, 29, 2503. (h) Grugel, H.; Minuth, T.; Boysen, M. M. K. Synthesis 2010, 3248. (i) Wang, Y.-Z.; Hu, X.-C.; Du, H.-F. Org. Lett. 2010, 12, 5482.

acid (3) or that of its derivatives to α,β -unsaturated ketones, 6,7 which have recently been under active investigation,⁸ are extremely promising. However, there has been only one report of asymmetric catalysis for the addition of the aryl group into 2,9,10 and the catalysis using 5% Rh/chiral bis-sulfoxide required the use of Ar₄BNa instead of 3 and a long reaction time of 24 h to give a moderate vield of 1. One of the reasons for this insufficient result is due to the less reactive substrate of 2 in the Rh-catalyzed catalysis. Recently, we reported a high-performance Rh catalyst bearing (6,6'-dimethoxybiphenyl-2,2'-diyl)bis[bis-(3,4,5-trifluorophenyl)phosphine] (MeO-F₁₂-BIPHEP, 4) for the asymmetric 1,4-addition of 3.¹¹ The reaction of 3 with an α,β -unsaturated ketone in the presence of Rh/(R)-4 showed a high turnover frequency of up to 53000 h⁻¹ in the model reaction 11b or gave optically pure 4-phenylchroman-2-one in high yield. In particular, in the latter case, the fact that the Rh/(R)-4 catalyst was highly effective for the less reactive coumarin substrate, which is a structural isomer of 2, raised our hopes for the success of the reaction of 2. Therefore, we attempted the efficient synthesis of chiral flavanones by using the Rh/4 catalyst.

Typically, the Rh-catalyzed asymmetric 1,4-additions have been performed using Rh-Cl species as precatalysts, which are converted to Rh-OH species as an active species by treatment with base such as KOH in situ. 12 However, flavanone 1 undergoes ring-opening under strongly basic conditions. 13 Therefore, we used a [RhOH(cod)]₂ complex as a catalyst precursor to avoid basic conditions. Although the use of [RhOH(cod)]₂ with chiral ligands for the asymmetric 1,4-addition often decreases the enantioselectivity of the products because of the high catalytic activity of [RhOH(cod)]₂ itself, ¹⁴ the reaction of **2a** with 2 equiv of phenylboronic acid (3a) in the presence of 1.5% [RhOH(cod)]₂ complex (3% Rh) in toluene/H₂O at 20 °C for 1 h gave no product (see Table 1, entry 1). 15 The addition of 1 equiv of (S)-4 for Rh largely accelerated the catalysis to yield 60% of (S)-flavanone (1aa) with 99% ee (entry 2). In this case, no remaining 3a was observed at the end of the reaction. When 3.5 equiv of 3a was used, a 95% yield of (S)-1aa was obtained (entry 3). The catalyst loading could be decreased to 0.5% Rh to yield 93% of (S)-1aa with 99% ee within just 1 h (entry 4). The effectiveness of the ligand (S)-4 was obvious because the reaction using 3% Rh/(S)-MeO-F₂₈-BIPHEP^{11a} or Rh/(S)-BINAP gave no product under the same conditions (entry 5 or 6).

Table 1. Asymmetric 1,4-Addition of 3a to 2a

entry	\mathbf{L}^*	Rh (%)	3a (equiv)	$\mathbf{1aa}^{a}\left(\% ight)$	ee ^b (%)
1	none	3	2.0	0	
2	(S)-4	3	2.0	60	99
3	(S)-4	3	3.5	95	99
4	(S)-4	0.5	3.5	93	99
5	$(S)\text{-}{\rm MeO}\text{-}{\rm F}_{28}\text{-}{\rm BIPHEP}$	3	2.0	0	
6	(S)-BINAP	3	2.0	0	

^a Isolated yield. ^b The enantiomeric excess was determined by chiral HPLC analyses (see the Supporting Information).

In the case of the reaction using Rh/(S)-4 (entries 3 and 4), 3% of the byproduct 2,4-diphenyl-4-chromanol (**5aaa**), ¹⁶ which was the 1,2- and 1,4-adduct, ¹⁷ was obtained as a single diastereomer. Because no **5aaa** was observed in the reaction at a 60% conversion (entry 2), the 1,2-addition of the phenyl group proceeded after 1,4-addition. Each phenyl group of **5aaa** was the anti relationship, which was determined by NOE (NOESY) experiment and DFT calculations (see the Supporting Information). Because the stereochemistry of the byproduct **5aaa** was identical to that of the product of the 1,2-addition of **3a** to (S)-**1aa** in the presence of 3% Rh/(R)-4 (eq 1), the absolute configuration of the product was (S,4R).

(S)-1aa + 3a
$$(4/Rh = 1.0)$$
 toluene / H₂O $(2S,4R)$ -5aaa 48%, 99% ee (1)

Next, we applied 4-tolylboronic acid (**3b**) to the asymmetric 1,4-addition of **2a** (Table 2). In this case, the formation of the byproduct **5abb** severely disturbed the synthesis of (S)-**1ab**. The reaction of **2a** with 3.5 equiv of **3b** in the presence of 0.5% [RhOH(cod)]₂ complex (1% Rh)

Org. Lett., Vol. 13, No. 8, **2011**

⁽⁹⁾ Chen, J.; Chen, J.-M.; Lang, F.; Zhang, X.-Y.; Cun, L.-F.; Zhu, J.; Deng, J.-G.; Liao, J. J. Am. Chem. Soc. 2010, 132, 4552.

⁽¹⁰⁾ Hoveyda reported Cu-catalyzed asymmetric addition of an alkyl group to **2**: Brown, M. K.; Degrado, S. J.; Hoveyda, A. H. *Angew. Chem., Int. Ed.* **2005**, *44*, 5306.

^{(11) (}a) Korenaga, T.; Osaki, K.; Maenishi, R.; Sakai, T. *Org. Lett.* **2009**, *11*, 2325. (b) Korenaga, T.; Maenishi, R.; Hayashi, K.; Sakai, T. *Adv. Synth. Catal.* **2010**, *352*, 3247. (c) Korenaga, T.; Maenishi, R.; Osaki, K.; Sakai, T. *Heterocycles* **2010**, *80*, 157.

⁽¹²⁾ Hayashi, T.; Takahashi, M.; Takaya, Y.; Ogasawara, M. J. Am. Chem. Soc. 2002, 124, 5052.

⁽¹³⁾ Cisak, A.; Mielczarek, C. J. Chem. Soc., Perkin Trans. 2 1992, 1603.

⁽¹⁴⁾ Lukin, K.; Zhang, Q.; Leanna, M. R. J. Org. Chem. 2009, 74, 929.

⁽¹⁵⁾ When the catalysis was perfomed at 50 °C for 6 h, (±)-1aa was obtained in 7% yield.

⁽¹⁶⁾ The absolute configuration had been unclear in the past report: Elkaschef, M. A. F.; Nosseir, M. H.; Mohamed, H.-E.-D. M. *J. Chem. Soc.* **1965**, 494.

^{(17) (}a) Vandyck, K.; Matthys, B.; Willen, M.; Robeyns, K.; Van Meervelt, L.; Van der Eycken, J. *Org. Lett.* **2006**, *8*, 363. (b) Iuliano, A.; Facchetti, S.; Funaioli, T. *Chem. Commun* **2009**, 457.

Table 2. Asymmetric 1,4-Addition of 3b to 2a

$$(Ar = 4-tol)$$

[RhOH(cod)]₂
 $/(S)-4$

O

HO

Ar

A

A

(2S,4R)-5abb

entry	Rh(%)	3b (equiv)	solvent	temp (°C)	time (h)	$\mathbf{2a}^{a}\left(\% ight)$	$\mathbf{1ab}^{a}\left(\% ight)$	$\mathbf{5abb}^{a}\left(\% ight)$
1	1	3.5	toluene	20	1	6	59	35
2	1	3.5	toluene	20	0.5	12	66	22
3	1	3.5	toluene	0	1	57	35	8
4	1	2.0	toluene	20	1	37	57	6
5	3	3.5	dioxane	20	1	72	28	0
6	3	3.5	dioxane	20	6	34	51	15
7	3	3.5	$ClCH_2CH_2Cl$	20	1	58	42	<1
8	3	3.5	$\mathrm{CH_2Cl_2}$	20	1	32	67	1
9	3	5.0	$\mathrm{CH_2Cl_2}$	20	1	8	82^b	10^c

^aThe ratios were determined by ¹H NMR. ^b Isolated yield was 82% with 99% ee. ^c Isolated yield was 7% with 99% ee.

with (S)-4 at 20 °C for 1 h proceeded to yield a 94% conversion, but the ratio of (S)-1ab/(2S,4R)-5abb was 59:35 (entry 1). At half the reaction time, 22% of (2S,4R)-**5abb** was observed (entry 2). When the reaction was performed under milder conditions, i.e., a lower temperature or a smaller amount of 3b, the byproduct (2S,4R)-5abb was already generated although the reaction proceeded moderately (entry 3 or 4). The toluene solvent greatly increased the catalytic activity of Rh/(S)-4, 11 resulting in a higher rate of 1,2-addition to (S)-1ab. Therefore, the other solvent was used to suppress the 1,2addition by deactivating the catalytic activity of Rh/(S)-4. The reaction in 1,4-dioxane in the presence of 3% Rh/(S)-4 for 1 h was very slow (28% conversion, entry 5). Although the conversion was improved to 66% by increasing the reaction time to 6 h, 15% of (2S,4R)-5abb was generated. The halogenated solvent afforded a better result. The reactions in 1,2-dichloroethane or dichloromethane for 1 h gave moderate yields of (S)-1ab with small amounts of (2S,4R)-**5abb** (entry 7 or 8). When the reaction with 5.0 equiv of 3b was performed, (S)-1ab was obtained in an 82% yield, and the formation of (2S,4R)-5abb was suppressed to 10% (entry 9). Although the catalytic activity of Rh/(S)-4 decreased as compared with that in toluene (Table 1), nevertheless, it was critically higher than that of previous reports.9

Several reactions of arylboronic acid 3 and substituted chromone 2 were performed in dichloromethane in the presence of 1.5% [RhOH(cod)]₂ complex (3% Rh) (Table 3). Each of these reactions required the optimization of the reaction conditions (i.e., the amount of 3, the reaction temperature, and the reaction time) to suppress the formation of the byproduct 5. Overall, the reactions gave almost optically pure products (S)-1 in yields of over 80% except for the reaction of 7-hydroxychromone (2f), which did not dissolve in the solvent (entry 11). When the hydroxy group

was protected with MOM, the reaction proceeded smoothly to yield 90% of (S)-1ga (entry 12). The product (S)-1ga could be deprotected by treatment with concd HCl in CH₂Cl₂ to yield 88% of (S)-1fa without loss of optical purity (99% ee).

Table 3. Asymmetric 1,4-Addition of 3 to 2

entry	2	3 [equiv]	temp (°C)	$1 (\%)^a [\% \text{ ee}]^b$	5 $(\%)^a$ $[\% \text{ ee}]^b$
1^f	2a	3b [5]	20	1ab, 82 [99]	5abb , 7 [>99]
2^f	2a	3c [3]	20	1ac , ^c 90 [>99]	$\mathbf{5acc},^{d} 2 [99]$
3^g	2a	3d [12]	20	1ad , ^c 83 [>99]	5add , 0
4^h	2a	3e [3]	0	1ae, 83 [99]	5aee , 3 [99]
5^g	2a	3f [15]	10	1af , ^c 80 [>99]	5aff , ^d 2 [>99]
6^e , g	2a	3g [13]	40	$\mathbf{1ag},^{c}80[99]$	$5agg,^d 6[97]$
7^i	2 b	3a [5]	40	1ba ,86 [>99]	5baa ,8 [98]
8^{j}	2c	3a [10]	35	1ca, 81 [>99]	5caa , 3 [>99]
9^g	2d	3a [8]	30	1da, 87 [>99]	5daa , 6 [>99]
10^i	2e	3a [8]	40	1ea , ^c 90 [>99]	$5eaa$, d 8 [94]
11^g	2f	3a [8]	30	1fa , 0	5faa , 0
12^g	2g	3a [8]	30	1ga , 91 [99]	5gaa , 0

^a Isolated yield. ^b The enantiomeric excess was determined by chiral HPLC analyses (see the Supporting Information). ^c Absolute configuration was unclear. In all compounds, negative optical rotation was observed. ^d Absolute configuration was unclear ((2S*,4R*)-form). ^e In toluene. ^f For 1 h. ^g For 2 h. ^h For 0.7 h. ⁱ For 1.5 h. ^j For 3 h.

2024 Org. Lett., Vol. 13, No. 8, 2011

This catalytic system was applicable to the synthesis of (S)-(-)-pinostrobin (6), which shows many biological activities including antiviral, 18 antioxidant, 19 antileukemic, 20 and antiaromatase²¹ activities, is an inhibitor of sodium channel²² and Ca²⁺ signaling pathways,²³ and inhibits intestinal smooth muscle contractions.²⁴ Although pinostrobin is an attractive natural product, its procurement almost always relies on extraction from a natural source.²⁵ There is only one method for the chemical synthesis of optically pure 6 through the intramolecular Mitsunobu reaction of substituted chiral 2-hydroxyphenyl propanone, which was synthesized from a chiral building block. 26 Meanwhile, no efficient synthetic procedure through an asymmetric catalysis has been reported. Therefore, we attempted the synthesis of optically pure 6 through Rh-catalyzed asymmetric 1,4-addition by using Rh/(S)-4 (Scheme 1). Substrate 2h was easily synthesized from commercially available 4',6'dimethoxy-2'-hydroxyacetophenone.²⁷ The asymmetric 1,4addition to 2h with 10 equiv of 3a was performed using 3% Rh/(S)-4 in CH_2Cl_2 at 40 °C for 2 h. However, 2h was a less reactive substrate, and the product 1ha was obtained in low yield (33%). Fortunately, the reaction in toluene yielded 92% of (S)-1ha (99.4% ee) without the generation of the 1,2-/ 1,4-adduct because of the steric repulsion of the o-methoxy group for the coordination of (S)-1ha to Rh. The reaction proceeded in the presence of 1% Rh/(S)-4 at 40 °C for 4 h to give 94% of (S)-1ha. When the reaction was performed at 60 °C for 3 h, the catalyst loading could be decreased to 0.5% to yield 93% of (S)-1ha with 99.6% ee. Furthermore, the regioselective demethylation of (S)-1ha with AlCl₃ was performed according to the published method²⁶ to yield 90% of (S)-(-)-6 in a natural form without loss of optical purity (99.6% ee). As a results, (S)-(-)-6 was synthesized in three steps with an overall yield of 74% from commercially available 4',6'-dimethoxy-2'-hydroxyacetophenone. The unnatural (R)-(+)- $\mathbf{6}$ could be prepared using the (R)- $\mathbf{4}$ ligand in the catalysis of $\mathbf{2h}$.

Scheme 1. Synthesis of Optically Pure Pinostrobin (6)

Rh	temp	3a	time	1ha	% ee	
3%	40 °C	10 equiv	1 h	92%	99.4	
1%	40 °C	12 equiv	4 h	94%	99.5	
0.5%	60 °C	8 equiv	3 h	93%	99.6	

In summary, we have developed a method for the highly enantioselective synthesis of flavanones through the Rhcatalyzed asymmetric 1,4-addition. This method promises to make the preparation of bioactive chiral flavanones easy from commercially available chromones or arylboronic acids. As a typical example, (S)- and (R)-pinostrobin could be efficiently synthesized through asymmetric catalysis. The synthesis was achieved using the rhodium catalyst with highly electron-poor chiral diphosphine MeO-F₁₂-BIPHEP. This catalyst would be widely applicable to the synthesis of other bioactive chiral compounds through the asymmetric 1,4-addition.

Acknowledgment. This work was patially supported by the Okayama Foundation for Science and Technology. We thank the SC-NMR Laboratory of Okayama University for ¹H, ¹³C, and ¹⁹F NMR measurements.

Supporting Information Available. Experimental procedures, spectroscopic data, copies of NMR spectra and HPLC, and confirmation of relative conformation of 5aaa by NOE (NOESY) experiments and DFT calculations. This material is available free of charge via the Internet at http://pubs.acs.org.

Org. Lett., Vol. 13, No. 8, 2011

⁽¹⁸⁾ Wu, N.; Kong, Y.; Zu, Y.; Fu, Y.; Liu, Z.; Meng, R.; Liu, X.; Efferth, T. *Phytomedicine* **2011**, *18*, 110.

⁽¹⁹⁾ Kong, Y.; Fu, Y.-J.; Zu, Y.-G.; Liu, W.; Wang, W.; Hua, X.; Yang, M. Food Chem. **2009**, 117, 152.

⁽²⁰⁾ Smolarz, H. D.; Mendyk, E.; Bogucka-Kocka, A.; Kocki, J. Z. Naturforsch. C: J. Biosci. 2006, 61, 64.

^{(21) (}a) Le Bail, J.-C.; Aubourg, L.; Habrioux, G. *Cancer Lett.* **2000**, *156*, 37. (b) Joshi, S. C.; Strauss, L.; Makela, S.; Santti, R. *J. Med. Food* **1999**, *2*, 235.

⁽²²⁾ Nicholson, R. A.; David, L. S.; Pan, R.-L.; Liu, X.-M. Fitoterapia 2010. 81. 826.

⁽²³⁾ Wangkangwan, W.; Boonkerd, S.; Chavasiri, W.; Sukapirom, K.; Pattanapanyasat, K.; Kongkathip, N.; Miyakawa, T.; Yompakdee, C. *Biosci. Biotech. Biochem.* **2009**, *73*, 1679.

⁽²⁴⁾ Meckes, M.; Paz, D.; Acosta, J.; Mata, R. Phytomedicine 1998, 5, 459.

^{(25) (}a) Kong, Y.; Zu, Y.-G.; Fu, Yu-J.; Liu, W. J. Food Compos. Anal. 2010, 23, 382. (b) Lutskii, V. I.; Tyukavkina, N. A.; Shostakovskii, M. F. Khim. Prir. Soedin. 1968, 4, 383. (c) Mongkolsuk, Stang; Dean, F. M. J. Chem. Soc. 1964, 4654. (d) Lindstedt, G. Acta Chem. Scand. 1950, 4, 1042. (e) Erdtman, H. Sven. Kem. Tidskr. 1944, 56, 2.

^{(26) (}a) Hodgetts, K. J. Tetrahedron 2005, 61, 6860. (b) Hodgetts, K. J. Tetrahedron Lett. 2001, 42, 3763.

⁽²⁷⁾ Oonuma, H.; Nishizama, Y.; Jokura, H. JP 05,301,813, 1993.